

DIARYLPROPANOIDS FROM *IRYANTHERA POLYNEURA**

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Abstract—The trunk wood of *Iryanthera polyneura* Ducke (Myristicaceae) contains pinocembrin, 1-(2',4'-dihydroxyphenyl)-3-(3'',4''-methylenedioxyphenyl)-propane, 1-(2',4'-dihydroxy-3'-methylphenyl)-3-(2''-methoxy-4'',5''-methylenedioxyphenyl)-propane and 4,2',4'-trihydroxy-3-methoxydihydrochalcone.

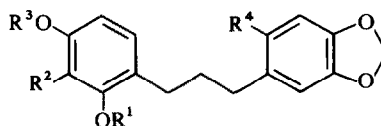
INTRODUCTION

The nomadic and very primitive Maku Indians of the Apaporis basin (Amazonia) employ a plaster made of the crushed leaves of *Iryanthera polyneura* Ducke (Myristicaceae) for treating seriously infected wounds and cuts [3]. This ethnopharmacological observation stimulated interest in the chemistry of the tree, and a study on the composition of its trunk wood was undertaken. The results are reported in the present paper.

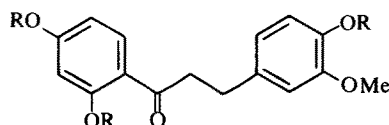
RESULTS

Fractionation of a benzene extract of the trunk wood afforded, besides sitosterol and pinocembrin, the 1,3-diarylpropane **1a**, previously obtained from *I. coriacea* Ducke [4], two novel compounds, the additional diarylpropane **1c** and the dihydrochalcone **2a**, and a small quantity of a mixture containing two chalcones tentatively formulated **3a** and **3b**.

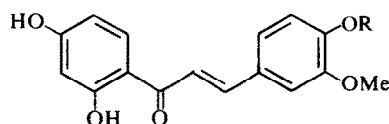
The basic structure of the diarylpropane, $C_{15}H_{10}O_2$ Me(OH)₂ OMe.O₂CH₂, was indicated by ¹H and ¹³C NMR spectral evidence which pointed to a CH₂.CH₂.CH₂ chain flanked by two aryl groups. One must bear 2-methoxy-4,5-methylenedioxy substitutions. The base peak in the MS shows the correct mass for a methoxy-methylenedioxytropylium ion; the ¹H NMR spectrum includes two singlets (δ 3.34 and 3.42). The other aryl group bears 2,4-dihydroxy-3-methyl substitution. The ¹H NMR spectrum includes two doublets (δ 6.36 and 6.72, *J* = 8 Hz); oxidation of the compound does not lead to *ortho* or *para*-quinones; a ¹³C NMR quartet at high field (δ 8.9) is compatible only with a methyl between two vicinal substituents [5]. The ¹³C NMR assignments for this



- 1a** R¹ = R² = R³ = R⁴ = H
1b R¹ = R² = R⁴ = H, R³ = Me
1c R¹ = R³ = H, R² = Me, R⁴ = OMe
1d R¹ = R³ = OAc, R² = Me, R⁴ = OMe



- 2a** R = H
2b R = Ac



- 3a** R = H
3b R = Me

1,3-diarylpropane (**2c**) are compatible with the analogous data for **1a** [4] and virolane (**1b**) (see Experimental), previously obtained from *Virola multinervia* Ducke [6].

The basic structure of the dihydrochalcone, $C_{15}H_{10}O_2$ (OH)₃ OMe, was indicated by ¹H and ¹³C NMR spectral evidence which pointed to a CH₂.CH₂ group flanked by benzoyl and aryl substituents. The benzoyl group must be 2,4-dihydroxylated: the molecule has chelated and acidic hydroxyls evidenced by IR and UV (AlCl₃ and NaOAc shifts) spectra; acetylation produces NMR shifts of the aromatic C signals which are closely comparable with the values observed for 2,4-dihydroxyacetophenone (see Experimental). The aryl group must bear 4-hydroxy-3-methoxy substitution: the two oxy-functions are located at C-3 and C-4 (H-2, H-5 and H-6 give a NMR multiplet at δ 6.7–6.9); acetylation produces paramagnetic shifts

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(+5.1 to +7.6 ppm) of the C-1, C-3 and C-5 signals, and a small diamagnetic shift (−2.1 ppm) of the C-4 signal, leaving the C-2 and C-6 signals practically unaffected. The MS of this dihydrochalcone (**2a**) is accordingly very simple, since both the dihydroxybenzoyl ion and the hydroxy-methoxytropylium ion have *m/e* 137.

The constituents of a mixture obtained in minute amount were tentatively characterized as chalcones **3a** and **3b** by ^1H NMR spectroscopy. Not only the molecular peaks 286 (55%) and 300 (45%) of the MS, but also the fragment peaks due to dihydroxybenzoyl (*m/e* 137 (100%)), hydroxy-methoxystyryl (*m/e* 150 (80%)) and dimethoxystyryl (*m/e* 164 (50%)) ions, were consistent with this hypothesis.

DISCUSSION

The biosynthesis of cinnamylphenols in Leguminosae was postulated to involve the cinnamylation of triacetate moieties [7]. The fact that both the Leguminosae and the Myristicaceae are isoflavonoid accumulators invites the hypothesis that, as a further common biochemical feature, cinnamylation also occurs in both families, the 1,3-diarylpropanes of Myristicaceae being simply hydrogenation products of intermediate cinnamylphenols. It has, nevertheless, been suggested recently that the co-occurrence of 1,3-diarylpropanes with 1,3-diarylpropan-2-ols and flavans points to their biosynthetic derivation from chalcones [8], a postulate which receives now additional support by the isolation of a flavanone and, most importantly, by the co-occurrence of chalcone-dihydrochalcone-1,3-diarylpropane with identical oxygenation pattern in *Iryanthera polyneura*.

EXPERIMENTAL

Isolation of the constituents. Trunk wood of *Iryanthera polyneura* was collected near Manaus, Amazonas State, and identified by the botanist Dr. William A. Rodrigues. A C_6H_6 extract (4%, 200 g) was chromatographed on a Si gel column. Petrol, $\text{C}_6\text{H}_6\text{--CHCl}_3$ and MeOH eluted, in order, aliphatic (100 g), aromatic (40 mg) and aliphatic (50 g) fractions. The aromatic fraction (25 g) was rechromatographed on a Si gel column. The following solvents eluted, in order, the indicated compounds C_6H_6 : **1c** (150 mg) and pinocembrin (120 mg), $\text{C}_6\text{H}_6\text{--CHCl}_3$ 8:2: sitosterol (500 mg), $\text{C}_6\text{H}_6\text{--CHCl}_3$ 1:1: **1a** (1 g), a mixture of **3a** and **3b** (8 mg after purification by aq. Na_2CO_3) and **2a** (160 mg).

1-(2',4'-Dihydroxy-3'-methylphenyl)-3-(2''-methoxy-4'',5''-methylenedioxyphenyl)-propane (**1c**). Mp 104–106° ($\text{C}_6\text{H}_6\text{--petrol}$); $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3500, 3400, 1620, 1190, 1160, 1060, 1030; $\lambda_{\text{max}}^{\text{EtOH}}$ nm: 227, 294 inf., 310 (ϵ 20200, 6000, 8200), no NaOAc, NaOAc + H_3BO_3 , AlCl_3 shifts, $\lambda_{\text{max}}^{\text{EtOH} + \text{NaOH}}$ nm: 253, 310 (ϵ 14200, 3100); Gibbs test: positive. ^1H NMR (100 MHz, $(\text{CD}_3)_2\text{CO}$): δ 1.65–1.95 (m, 2H-2), 2.13 (s, Me), 2.56 and 2.58 (2t, J = 6 Hz, 2H-1 and 2H-3), 3.74 (s, OMe), 5.86 (s, O_2CH_2), 6.36 (d, J = 8 Hz, H-5'), 6.58 (s, H-3''), 6.66 (s, H-6''), 6.72 (d, J = 8 Hz, H-6'), 6.96 and 7.81 (2s, 2OH). MS (*m/e*): 316 (48%) M^+ , 165 (100), 151 (13), 137 (28). ^{13}C NMR **1a/1b/1c** (25.2 MHz,

$(\text{CD}_3)_2\text{CO}$): δ —/—/8.9 (q, Me), 29.6/28.6/31.5 (t, C-2), 32.7/31.7/30.5 (t, C-1), 35.7/35.2/30.4 (t, C-3), —/56.2/56.5 (q, OMe), 101.1/101.6/101.4 (t, O_2CH_2), 103.1/100.6/120.4 (d/d/s, C-3'), 106.9/105.7/107.1 (d, C-5'), 108.3/108.0/109.9 (d, C-6'), 109.1/108.3/95.3 (d, C-3''), 120.0/120.3/123.7 (s, C-1'), 121.5/121.0/152.9 (d, C-2''), 130.8/130.4/127.1 (d, C-6'), 137.0/136.0/123.7 (s, C-1''), 147.9/147.3/141.4 (s, C-5''), 148.0/145.3/146.6 (s, C-4'), 156.1/154.1/154.1 (s, C-2'), 156.7/158.3/154.6 (s, C-4') (for the sake of comparison, the C of **1a** and **1b** substituted by R^4 = H, see formula, was also numbered C-2''). Diacetate (**1d**). ^1H NMR (100 MHz, CDCl_3): δ 1.6–2.0 (m, 2H-2), 1.96 (s, Me), 2.26 and 2.30 (2s, 2OAc), 2.45–2.65 (m, 2H-1 and 2H-3), 3.74 (s, OMe), 5.87 (s, O_2CH_2), 6.50 (s, H-3''), 6.63 (s, H-6''), 6.87 (d, J = 8 Hz, H-5'), 7.08 (d, J = 8 Hz, H-6').

4,2',4'-Trihydroxy-3-methoxydihydrochalcone (**2a**). Mp 153–155° ($\text{C}_6\text{H}_6\text{--petrol}$); $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3460, 3350, 1625; $\lambda_{\text{max}}^{\text{EtOH}}$ nm: 218, 232 inf., 280, 318 (ϵ 18400, 12900, 15200, 7200), slight NaOAc and AlCl_3 shifts, no NaOAc + H_3BO_3 shift, $\lambda_{\text{max}}^{\text{EtOH} + \text{NaOH}}$ nm: 227, 249, 342 (ϵ 7200, 12800, 28800), Gibbs test: positive. ^1H NMR (60 MHz, CDCl_3): δ 2.9–3.2 (m, 2CH₂), 3.90 (s, OMe), 6.40 (dd, J = 8, 2 Hz, H-5'), 6.42 (H-3'), 6.7–6.9 (m, H-2, 5, 6), 7.80 (d, J = 8 Hz, H-6'), 12.7 (OH-2'). MS (*m/e*): 288 (58%) M^+ , 151 (12), 150 (21), 137 (100). Triacetate (**2b**). oil. ^1H NMR (60 MHz, CDCl_3): δ 2.3 (s, 3OAc), 3.0–3.3 (m, 2CH₂), 3.80 (s, OMe), 6.85 (H-2), 7.0 (H-3'), 6.85–7.05 (H-5,6), 7.05 (dd, J = 8, 2 Hz, H-5'), 7.80 (d, J = 8 Hz, H-6'). ^{13}C NMR **2a/2b** (20 MHz, CDCl_3): δ —/20.6 (s, OAc), —/21.0 (s, 2OAc), 30.8/29.9 (t, CH₂— β), 40.0/43.0 (t, CH₂— α), 56.1/55.9 (q, OMe), 103.5/119.1 (d, C-3'), 108.4/117.4 (d, C-5'), 112.1/112.9 (d, C-2), 113.5/128.0 (s, C-1'), 115.1/122.7 (d, C-5), 121.1/120.4 (d, C-6), 132.5/130.7 (d, C-6'), 133.0/138.2 (s, C-1), 142.1/140.0 (s, C-4), 144.8/149.9 (s, C-3), 164.8/154.0 (s, C-4'), 165.7/151.1 (s, C-2'), —/168.3 (s, CO), —/169.0 (s, 2CO), 206.8/197.9 (s, CO). ^{13}C NMR shifts **2a** → **2b**/2,4-dihydroxyacetophenone [9] → 2,4-diacetoxyacetophenone (ppm): +14.5/ +13.8 (C-1'), −14.6/ −14.9 (C-2'), +15.5/ +15.5 (C-3'), −10.8/ −11.4 (C-4'), +9.0/ +7.7 (C-5'), −1.8/ −3.8 (C-6').

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REFERENCES

- Barata, L. E. S., Baker, P. M., Gottlieb, O. R. and Ruveda, E. A. (1978) *Phytochemistry* **17**, 783.
- Braz Filho, R., Gottlieb, O. R., Moraes, A. A. de, Pedreira, G., Pinho, S. L. V., Magalhães, M. T. and Ribeiro, M. N. de S. (1977) *Lloydia* **40**, 236.
- Schultes, R. E. and Holmstedt, B. (1971) *Lloydia* **34**, 61.
- Lima, R. A. de, Franca, N. C., Diaz, D., P. P. and Gottlieb, O. R. (1975) *Phytochemistry* **14**, 1831.
- Stothers, J. B. (1972) *Carbon-13 NMR Spectroscopy*, p. 202. Academic Press, New York.
- Braz Filho, R., Leite, M. F. F. and Gottlieb, O. R. (1973) *Phytochemistry* **12**, 417.
- Ollis, W. D. and Gottlieb, O. R. (1968) *Chem. Commun.* 1396.
- Gottlieb, O. R. (1977) *Isr. J. Chem.* **16**, 45.
- Ternai, B. and Markham, K. R. (1976) *Tetrahedron* **32**, 565.